

Na Zuo and Hong-Wu He\*

Key Laboratory of Pesticides and Chemical Biology, College of Chemistry, Central China Normal University, Wuhan 430079, People's Republic of China

Correspondence e-mail:  
 he1208@mail.ccnu.edu.cn

**Key indicators**

Single-crystal X-ray study  
 T = 298 K  
 Mean  $\sigma(C-C) = 0.004 \text{ \AA}$   
 R factor = 0.051  
 wR factor = 0.140  
 Data-to-parameter ratio = 18.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

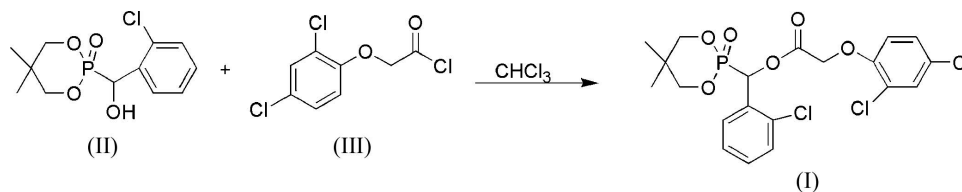
## 2-[(2-Chlorophenyl)[(2,4-dichlorophenoxy)-acetoxy]methyl]-5,5-dimethyl-1,3,2-dioxaphosphinan-2-one

In the title compound,  $C_{20}H_{20}Cl_3O_6P$ , all bond lengths and angles are normal. The P atom is in a distorted tetrahedral configuration. The crystal packing is stabilized by weak intermolecular C—H...O hydrogen bonds and  $\pi$ – $\pi$  interactions.

Received 9 January 2007  
 Accepted 16 January 2007

**Comment**

Some hydroxyphosphonates and their derivatives have shown good biological and pharmaceutical activities (Chen *et al.*, 1995). The title compound, (I), has been prepared as part of our work directed towards the synthesis of oxyalkylphosphonic acid derivatives exhibiting biological activities (He *et al.*, 2005).



Bond distances and angles in (I) are as expected, and the dioxaphosphorinane ring adopts a chair conformation (Fig. 1). The bond angles around atom P1 (Table 1) suggest that its environment adopts a distorted tetrahedral configuration, as observed previously in a similar compound (Zuo & He, 2006). The torsion angles C16–O4–C13–C14, O4–C13–C14–O6 and C14–O6–C15–C20 are  $-174.38$  (18),  $163.60$  (19) and  $-0.90$  (4)°, respectively.

Weak intermolecular C—H...O hydrogen bonds (Table 2) and  $\pi$ – $\pi$  interactions [ $Cg \cdots Cg^{iv} = 3.7939$  (18) Å, where Cg is the centroid of the C15–C20 ring; symmetry code: (i)  $1 - x$ ,  $1 - y$ ,  $1 - z$ ] stabilize the crystal packing.

**Experimental**

2-[Hydroxy(2-chlorophenyl)methyl]-5,5-dimethyl-1,3,2-dioxaphosphinan-2-one, (II) (Sudha *et al.*, 1997), and (2,4-dichlorophenoxy)acetyl chloride, (III) (Klaassens & Schoot, 1956), were prepared according to the literature procedures. Compound (II) (0.022 mol) in chloroform (10 ml) was added dropwise at 275–277 K to a stirred solution of (III) (0.02 mol) and triethylamine (0.028 mol) in chloroform (15 ml). The mixture was stirred at 283–288 K for 2–3 h, washed with 0.5% hydrochloric acid solution followed by saturated sodium hydrogen carbonate solution, dried and evaporated. Recrystallization from dichloromethane gave colourless blocks of (I).

## Crystal data

$C_{20}H_{20}Cl_3O_6P$   
 $M_r = 493.68$   
 Triclinic,  $P\bar{1}$   
 $a = 8.8627$  (7) Å  
 $b = 10.4972$  (9) Å  
 $c = 13.3624$  (11) Å  
 $\alpha = 79.8340$  (10)°  
 $\beta = 83.7070$  (10)°  
 $\gamma = 69.9360$  (10)°

$V = 1147.75$  (16) Å<sup>3</sup>  
 $Z = 2$   
 $D_x = 1.428$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 $\mu = 0.50$  mm<sup>-1</sup>  
 $T = 298$  (2) K  
 Block, colourless  
 $0.30 \times 0.20 \times 0.20$  mm

## Data collection

Bruker SMART APEX CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: none  
 12883 measured reflections

4976 independent reflections  
 4228 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.024$   
 $\theta_{max} = 27.0^\circ$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.051$   
 $wR(F^2) = 0.140$   
 $S = 1.05$   
 4976 reflections  
 273 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0677P)^2 + 0.5055P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} = 0.001$   
 $\Delta\rho_{max} = 0.60$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.52$  e Å<sup>-3</sup>

Table 1

Selected bond angles (°).

O3–P1–O2	113.75 (10)	O3–P1–C6	112.63 (10)
O3–P1–O1	112.20 (10)	O2–P1–C6	105.53 (9)
O2–P1–O1	106.84 (8)	O1–P1–C6	105.26 (9)

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C17–H17 <sup>i</sup> ⋯O2 <sup>i</sup>	0.93	2.60	3.463 (3)	155
C20–H20 <sup>ii</sup> ⋯O3 <sup>ii</sup>	0.93	2.50	3.417 (3)	167
C14–H14A <sup>iii</sup> ⋯O3 <sup>iii</sup>	0.97	2.43	3.311 (3)	150
C8–H8 <sup>iii</sup> ⋯O4 <sup>iii</sup>	0.93	2.60	3.402 (3)	145
C5–H5A <sup>iii</sup> ⋯O5 <sup>iii</sup>	0.96	2.59	3.488 (4)	156

Symmetry codes: (i)  $x, y, z - 1$ ; (ii)  $-x + 1, -y + 1, -z + 2$ ; (iii)  $-x + 1, -y, -z + 2$ .

All H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C–H distances of

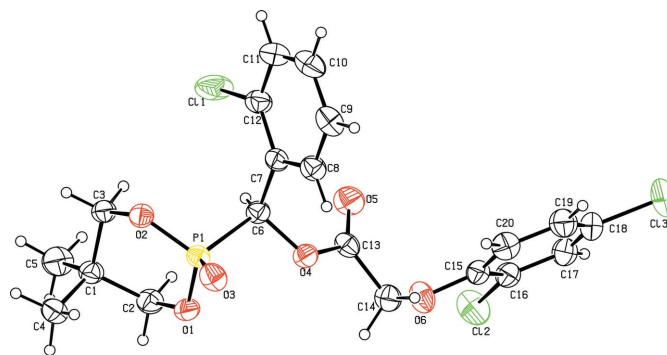


Figure 1

The molecular structure of (I), showing the atom-labelling scheme and with 50% probability displacement ellipsoids.

0.93–0.98 Å, and with  $U_{iso}(H) = 1.2U_{eq}(\text{aromatic C, CH, CH}_2)$  or  $1.5U_{eq}(\text{methyl C})$ .

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINTE* (Bruker, 2000); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

The authors gratefully acknowledge the financial support of this work by the National Basic Research Programme of China (grant No. 2003CB114400) and the National Natural Science Foundation of China (grant No. 20372023).

## References

- Bruker (1997). *SHELXTL*. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Bruker (2000). *SMART* (Version 5.059), *SAINTE* (Version 6.01) and *SADABS* (Version 6.10). Bruker AXS Inc., Madison, Wisconsin, USA.  
 Chen, R. Y., Mao, L. J. & Zhang, M. (1995). *Sci. China B*, **25**, 240–246.  
 He, H. W., Wang, T. & Yuan, J. L. (2005). *J. Organomet. Chem.* **690**, 2608–2613.  
 Klaassens, K. H. & Schoot, C. J. (1956). *Recl Trav. Chim. Pays-Bas*, **75**, 265–267.  
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.  
 Sudha, K., Senthamizh, R. S. & Kumara, K. C. S. (1997). *Synthesis*, pp. 207–211.  
 Zuo, N. & He, H.-W. (2006). *Acta Cryst.* **E62**, o4864–o4865.